

## Short, Stereoselective Syntheses of C(1→3)-linked Disaccharides

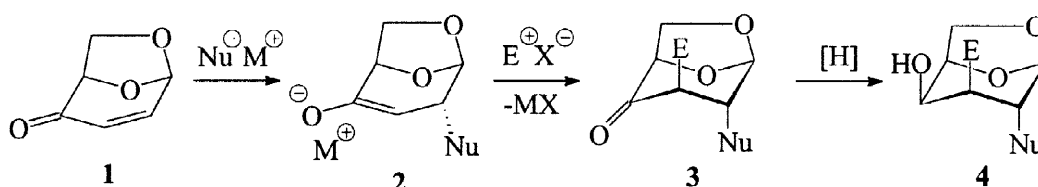
Yao-Hua Zhu and Pierre Vogel\*

Section de Chimie de l'Université de Lausanne, BCH,  
CH-1015 Lausanne-Dorigny, Switzerland

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**Abstract:** Nucleophilic additions to D-isolevoglucosenone are face selective and generate the corresponding enolates which react with 1,2:3,4-di-*O*-isopropylidene- $\alpha$ -D-galacto-hexodialdo-1,5-pyranose to give aldol adducts. These can be reduced stereoselectively into C(1→3) glycosides of 1,6-anhydro-D-galacto-pyranose. © 1997 Elsevier Science Ltd. All rights reserved.

The interaction between cell surface carbohydrates and their protein receptors is implicated in several important biological events.<sup>1</sup> Carbohydrate mimics are potentially useful tools to study cellular interactions and may represent leads for drug discovery.<sup>2</sup> In particular, C-linked disaccharides<sup>3</sup> offer the advantage of being resistant to acidic and enzymatic hydrolysis. They are therefore potential inhibitors of glycosidases and may represent non-hydrolysable epitopes. Since the first synthesis of a  $\beta$ -(1→6)-C-disaccharide by Rouzaud and Sinay<sup>4</sup> in 1983, several approaches to C-disaccharides have been proposed.<sup>3,5</sup> They are invariably multiple-step syntheses which do not always offer the necessary versatility for wide molecular diversity.<sup>6</sup> Hence, there is still a need for shorter and more convergent synthetic schemes. We report our latest efforts toward this goal which feature the conjugate additions of nucleophiles to isolevoglucosenone (**1**) followed by electrophilic quenching of the intermediate enolate **2** and reduction of the ketone **3** (*Scheme 1*).

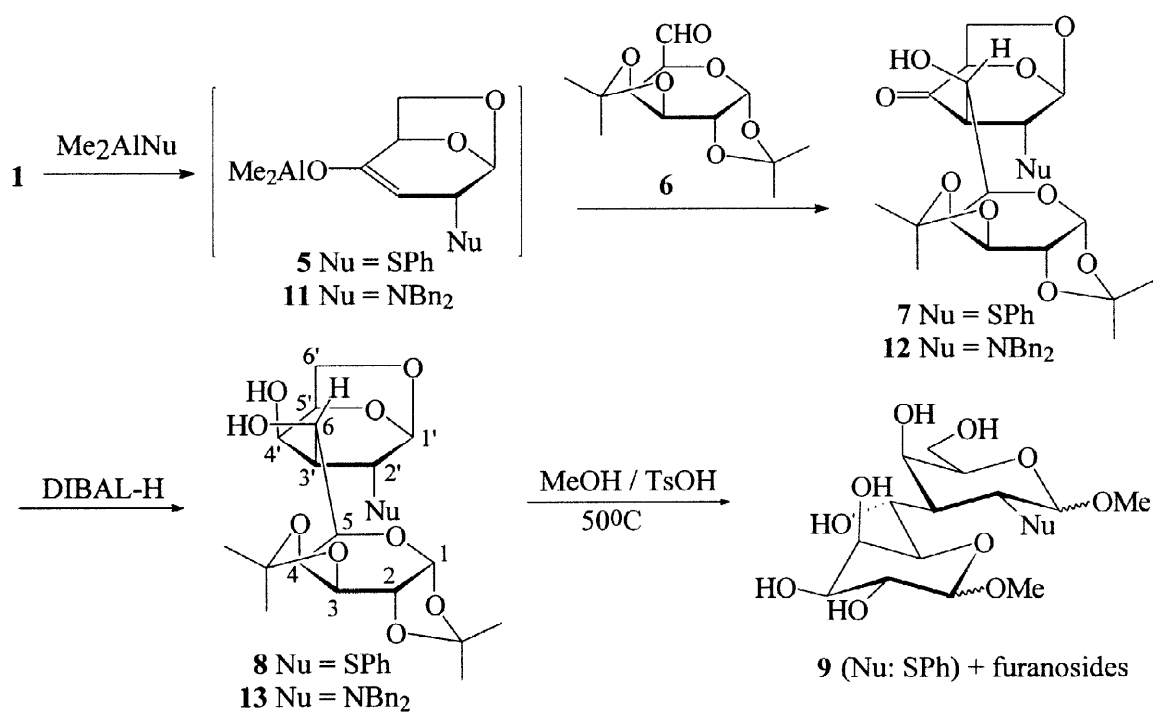


*Scheme 1*

Recently, Witczak and coworkers<sup>7</sup> used levoglucosenone (1,6-anhydro-3,4-dideoxy- $\beta$ -D-glycero-hex-3-enopyran-2-ulose) to generate C-disaccharides through Michael addition of ( $\beta$ -D-gluco-pyranosyl) nitromethane. This provided  $\beta$ -C(1→4)glucopyranosides of 3-deoxy-D-glucose. Isolevoglucosenone (**1**) is readily derived from D-glucose in four synthetic steps.<sup>8</sup> Enone **1** is known to add nucleophiles<sup>8,9</sup> and radicals<sup>10</sup> exclusively on its less hindered face (*syn* with respect to the oxa bridge). Oshima and co-workers<sup>11</sup> have shown that conjugate addition of Me<sub>2</sub>AlSPH to simple enones, followed by reaction of the aluminum enolates with aldehydes allow the preparation of the corresponding aldols in one-pot procedures. When a 1:1 mixture of **1**

\*e-mail: pierre.vogel@ico.unil.ch; fax: +41 21 692 39 75

and  $\text{Me}_2\text{AlSPh}$  was allowed to react at  $-78^\circ\text{C}$  in  $\text{CH}_2\text{Cl}_2$ , the corresponding aluminum enolate **5** was formed (Scheme 2). One equivalent of aldehyde **6** (derived in two steps from D-galactose<sup>12</sup>) was then added and allowed to react for 3h at  $-78^\circ\text{C}$ . A solution of DIBAL-H in  $\text{CH}_2\text{Cl}_2$  was then added dropwise and the reaction mixture was stirred at  $20^\circ\text{C}$  for 1h. This afforded the semi-protected C-disaccharide **8** which was isolated in 60% yield after flash chromatographic purification on silica gel. No other stereoisomeric product was isolated. Furthermore, the  $^1\text{H-NMR}$  spectrum of the crude reaction mixture did not reveal other stereomeric C-disaccharide. Acidic methanolysis ( $\text{MeOH}$ ,  $\text{TsOH}$ ,  $50^\circ\text{C}$ , 5h) led to a mixture of unprotected methyl pyranosides **9** and the corresponding methyl furanosides.

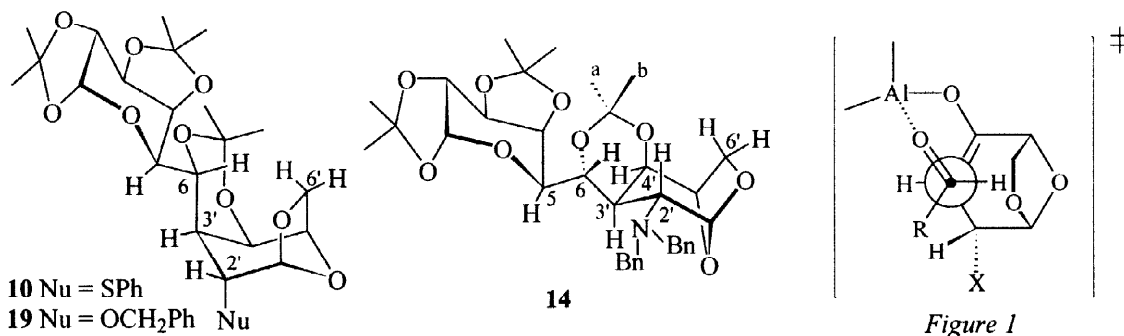


Scheme 2

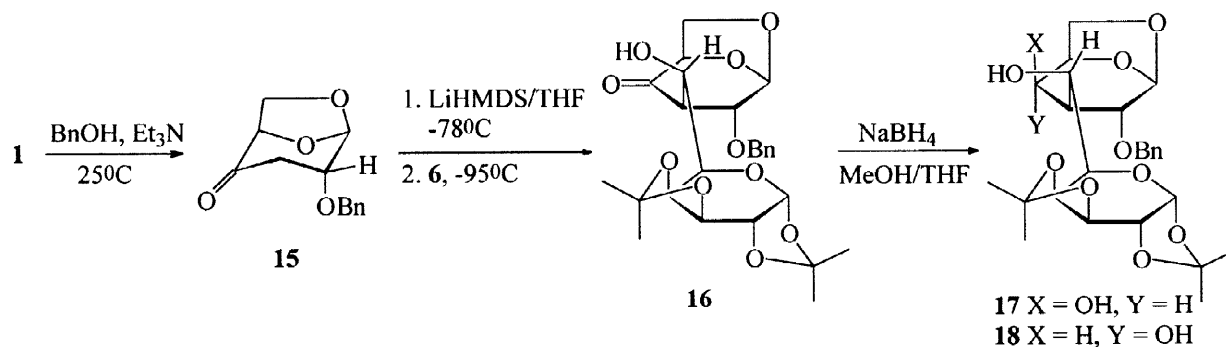
This result demonstrates that the doubly convergent approach depicted in Scheme 1 is not only feasible but is also highly stereoselective, the “aglycone” part of the C-glycoside having the configuration of D-galactose. The relative configuration of aldol **7** and of its reduction product **8**<sup>13</sup> was established by  $^1\text{H-NMR}$  (400 MHz, COSY, NOESY) of the acetonide **10**, obtained on treatment of **8** with a 3:1 mixture of acetone and 2,2-dimethoxypropane in the presence of  $\text{TsOH}$  and Drierite at  $20^\circ\text{C}$ . A coupling constant  $^3J(\text{H-3}', \text{H-6}) = 9.1$  Hz proves the *trans* relationship between these two protons. Strong NOE's observed between proton pairs H-2'/H-6 and H-6/H-6' confirm the *galacto* relative configuration of C(2',3',4',5'). These results demonstrate that the face *anti* with respect to the oxa bridge of enolate **5** is preferred for the cross-aldolisation, probably for steric reasons, the *exo* face being hindered by the substituent at C(2). The high diastereoselectivity of the cross-aldolisation can be interpreted in terms of the Zimmerman-Traxler model shown in Figure 1 (closed transition state, control by steric factors).<sup>14</sup>

Replacing  $\text{Me}_2\text{AlSPh}$  by  $\text{Me}_2\text{AlNBn}_2$ <sup>15</sup> allowed one to generate the corresponding C-glycoside of protected D-galactosamine **13** which was characterized as its diacetate<sup>16</sup>. The configuration was established by  $^1\text{H-NMR}$  (400 MHz, COSY, NOESY) of its acetonide **14**, prepared by the same way as acetonide **10**. Boat

conformation of the tetrahydropyran ring in the aglycone part results in a  $^3J(\text{H-2}', \text{H-3}') = 10.0$  Hz and NOE's between proton pairs H-2'/H-6 (strong), H-6/H-6' (weak), H-Me<sub>a</sub>/H-5, H-Me<sub>a</sub>/H-4', H-Me<sub>b</sub>/H-2' and H-Me<sub>b</sub>/H-6. Using Me<sub>2</sub>AlOBn<sup>17</sup> failed to produce the expected C-disaccharide with a 2-O-benzyl D-galactose. An alternative procedure had to be developed.



Conjugate addition of LiOBn or NaOBn to **1** at  $-50^\circ\text{C}$  produced the desired adduct **15** in low yield, the major product arising from the Michael addition of the resulting enolate to the starting enone. In the presence of Et<sub>3</sub>N, however, benzyl alcohol (as solvent) added to **1** at  $20^\circ\text{C}$  and afforded **15** in 90% yield. As expected, the reaction was highly stereoselective. The lithium enolate of **15** obtained on treatment with one equivalent of (Me<sub>3</sub>Si)<sub>2</sub>NLi at  $-78^\circ\text{C}$  (THF) did not eliminate lithium benzylate and could be reacted with all kinds of aldehydes, including **6**.<sup>18</sup> When the cross-aldolisation was carried out at  $-95^\circ\text{C}$ , with **6**, a single aldol **16** was obtained in 79% yield. Its reduction with NaBH<sub>4</sub> (MeOH/THF,  $0^\circ\text{C}$ ) was less stereoselective than reduction **7**→**8**, and gave a 5:1 mixture of **17**<sup>19</sup> and **18**. The major diol **17** has the configuration of D-galactose for its aglycone moiety as established by <sup>1</sup>H-NMR ( $^3J(\text{H-3}', \text{H-6}) = 10.3$  Hz, NOE's between proton pairs H-2'/H-6 and H-6/H-6') of its acetonide **19**, which was obtained on treating **17** (readily separated from **18** by flash chromatography on silica gel) with acetone/2,2-dimethoxypropane/TsOH/Drierite ( $25^\circ\text{C}$ ).



Our aldols should allow one to introduce fluoro or amino substituents at the methylene linker. Inversion at C-4' in our C-disaccharides should generate the corresponding D-glucose derivatives. Work is underway in our laboratory in order to evaluate the versatility of this new method of synthesis of C-disaccharides and analogues.

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## References and Notes:

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- Data for **8**: colorless oil,  $[\alpha]_D^{25} = -45$  (*c* 0.4, CHCl<sub>3</sub>); <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>): 7.49-7.43, 7.35-7.29, and 7.28-7.23 (*m*, 5H, Ph), 5.62 (*s*, H-1'), 5.54 (*d*, *J* = 5.1, H-1), 4.67 (*t*, *J* = 6.6, H-4'), 4.59 (*dd*, *J* = 6.6, 4.8, H-5'), 4.54 (*dd*, *J* = 7.9, 2.1, H-4), 4.52 (*dd*, *J* = 8.2, 2.7, H-3), 4.38 (*d*, *J* = 7.6, H<sub>a</sub>-6'), 4.32 (*dd*, *J* = 4.8, 2.4, H-2), 4.11 (*dd*, *J* = 7.9, 1.8, H-6), 3.94 (*dd*, *J* = 7.9, 1.8, H-5), 3.52 (*dd*, *J* = 7.6, 4.8, H<sub>b</sub>-6'), 3.38 (*d*, *J* = 7.9, H-2), 2.23 (*m*, H-3), 1.54, 1.46, 1.34, 1.32 (4*s*, Me × 4); <sup>13</sup>C-NMR (100.61 MHz, CDCl<sub>3</sub>): 134.1, 131.8, 129.2, 109.6, 108.8, 103.4, 96.6, 74.8, 71.3, 70.8, 70.3, 67.0, 66.9, 62.8, 49.1, 38.2, 25.9, 25.9, 24.9, 24.4.
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- Prepared by addition of Me<sub>3</sub>Al to a CH<sub>2</sub>Cl<sub>2</sub> solution of benzyl alcohol at 0°C.
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- Data for **17**: mp: 155-156°C;  $[\alpha]_D^{25} = -27$  (*c* 0.65, CHCl<sub>3</sub>); <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>): 7.43-7.29 (*m*, 5H), 5.56 (*d*, *J* = 5.1, H-1), 5.47 (*s*, H-1'), 4.72 and 4.53 (2*d*, *J* = 11.8), 4.67-4.58 (*m*, 3H), 4.42 (*dd*, *J* = 7.9, 1.8, H-4), 4.35 (*dd*, *J* = 5.1, 2.4, H-2), 4.34 (*d*, *J* = 7.6, H<sub>a</sub>-6'), 4.25 (*dd*, *J* = 8.5, 4.5, H-6), 3.98 (*dd*, *J* = 8.5, 1.8, H-5), 3.53 (*d*, *J* = 8.2, H-2'), 3.46 (*dd*, *J* = 7.3, 4.5, H<sub>b</sub>-6'), 2.18 (*m*, H-3'), 1.54, 1.49, 1.37, 1.34 (4*s*, Me × 4); <sup>13</sup>C-NMR (100.61 MHz, CDCl<sub>3</sub>): 137.9, 128.5, 127.9, 109.5, 108.7, 102.0, 96.5, 77.1, 73.9, 71.6, 71.0, 70.8, 70.4, 70.1, 67.6, 65.3, 61.9, 38.7, 26.0, 25.9, 24.9, 24.6.